

The environmentally-sound management of agricultural phosphorus

Andrew N. Sharpley & Paul J.A. Withers

*USDA-ARS, National Agricultural Water Quality Lab., P.O. Box 1430, Durant, OK 74702–1430 and ADAS
Bridgets Research Centre, Martyr Worthy, Winchester, Hampshire SO21 IAP, United Kingdom*

Received 23 December 1993; accepted in revised form 17 June 1994

Abstract

Freshwater eutrophication is often accelerated by increased phosphorus (P) inputs, a greater share of which now come from agricultural nonpoint sources than two decades ago. Maintenance of soil P at levels sufficient for crop needs is an essential part of sustainable agriculture. However, in areas of intensive crop and livestock production in Europe and the U.S.A., P has accumulated in soils to levels that are a long-term eutrophication rather than agronomic concern. Also, changes in land management in Europe and the U.S.A. have increased the potential for P loss in surface runoff and drainage. There is, thus, a need for information on how these factors influence the loss of P in agricultural runoff. The processes controlling the build-up of P in soil, its transport in surface and subsurface drainage in dissolved and particulate forms, and their biological availability in freshwater systems, are discussed in terms of environmentally sound P management. Such management will involve identifying P sources within watersheds; targeting cost-effective remedial measures to minimize P losses; and accounting for different water quality objectives within watersheds. The means by which this can be achieved are identified and include developing soil tests to determine the relative potential for P enrichment of agricultural runoff to occur; establishing threshold soil P levels which are of environmental concern; finding alternative uses for animal manures to decrease land area limitations for application; and adopting management systems integrating measures to reduce P sources as well as runoff and erosion potential.

Introduction

Since the 1970's point sources of water pollution have been to large extent controlled, due to their easy identification [50]. Even so, there are still water quality problems [31, 59, 96]. One of the current concerns is the accelerated eutrophication of surface waters from increased nutrient inputs stimulating algal and rooted aquatic plant growth. Controlled eutrophication is beneficial to the fisheries industry. However, excessive eutrophication restricts water use for recreation, industry, and drinking due to the increased growth of undesirable algae and aquatic weeds, which by their senescence and decay, can cause oxygen shortages and fish kills. In addition, potentially carcinogenic toxins produced by some blue-green algal blooms (dominantly cyanobacteria), can pose acute health risks to humans and animals if consumed [35, 37, 55]. These toxins

also contribute to unpalatability of drinking water via trihalomethane formation during water dechlorination [61]. Thus, eutrophication can create serious local and regional economic problems.

Nitrogen (N), carbon (C) and phosphorus (P) are the major nutrients required for freshwater eutrophication. However, most attention has focused on controlling P inputs, because of the free air-water exchange of N and C and fixation of atmospheric N by some blue-green algae. Irrespective of whether point sources (e.g., municipal and industrial discharge) are controlled or not, the nonpoint input of P in agricultural runoff can sustain further eutrophication of freshwaters in Europe [8, 23, 36, 41, 97, 98], North America [16, 40, 65, 67], and Oceania [100, 101, 102]. There is little reason to believe that these environmental concerns will be different in other regions of the world, where agricultural systems involve increased P inputs.

Current concerns facing the environmentally-sound management of P in agriculture are similar world-wide and revolve around agricultural, economic, and environmental compromises associated with balancing productivity with environmental values. For example, if agricultural P inputs are based on environmental rather than agronomic criteria, the generally lower inputs may compromise crop productivity [104]. Similarly, if manure applications are based on P rather than N input, the resulting lower rates of application will force many farmers to find alternative ways of disposing of excess manure and to buy needed inorganic N.

Phosphorus inputs are required for profitable crop production, especially in areas with P deficient soils. Thus, balancing P inputs and outputs is one of the main challenges facing modern farming systems which need to be both economically and environmentally sound. To meet these future challenges, we must be able to identify sites, soils or management systems that are vulnerable to P loss so that appropriate remedial measures can be effectively targeted. To achieve this, environmental assessments are required at the farm and watershed level; for example, soil tests are needed to assess a soil's potential to release P to runoff rather than its availability for plant uptake. Also, economically viable practices that minimize P loss in drainage and runoff should be identified and the means for their effective implementation developed. Effective implementation will involve education programs to overcome the common perception among end-users of water that it is often much cheaper to treat the symptoms of eutrophication rather than control the diffuse or nonpoint sources.

This paper discusses the major sources, forms, and pathways of P loss from agriculture in Europe and the U.S.A., how we can overcome current concerns and meet future challenges facing P management, and the development of sustainable farming systems that maintain agricultural productivity, minimize environmental degradation, and promote short- and long-term economic viability and stable farming communities.

Agricultural phosphorus

An assessment of the agricultural contribution to eutrophication in various European countries has been attempted by Vighi and Chiaudani [98]. This assessment, which was based upon a detailed questionnaire sent to participating countries, shows that agriculture

contributes between 24 and 71% of the total P (TP) loadings to surface waters in Europe, although these estimates include both point (10–53% of total P load) and nonpoint sources (8–30% of total P load). Point sources of P loss from agriculture include concentrated discharge of livestock manure, accidental discharge from manure stores or direct contamination during broadcast applications of fertilizer and manure. Such point sources are best controlled by appropriate legislation relating to codes of good agricultural practice [50]. The contribution of the diffuse sources of P from agricultural land is a function of the P loading to soil and land management practices which affect the ease with which accumulated P is lost.

In the Republic of Ireland, a balance sheet study of P inputs and outputs indicated that inputs are more than double outputs, although inorganic P fertilizer use has remained constant over the last 10 years [94]. The available P content of soils have consequently shown a steady, almost linear, eight fold increase since 1950. The positive balance represents the fertilization with inorganic P in excess of crop requirements, because the manurial value of P inputs from livestock are often ignored by the farming community. A large proportion (75%) of the TP ingested by livestock in concentrate and forage is excreted [1, 22]. The P excreted from housed livestock requires storage and is often applied to a relatively small land area.

In the U.K., although the consumption of inorganic fertilizer has remained fairly constant over the last 30 years and national soil surveys in England and Wales have indicated no substantial increase in available soil P over the period 1969–1988 [88], there is still a large percentage of arable (56%) and grassed fields (30%) with moderate (26 to 45 mg kg⁻¹) to high (> 45 mg kg⁻¹) Olsen extractable soil P levels (Table 1). In reviewing the significance of agriculture as a source of P to inland and coastal waters in the U.K., Withers [103] concluded that recent changes in land management practice (e.g., increase in winter cereals, slurry based livestock systems and land underdrainage) rather than changes in agricultural P inputs had increased the potential for P loss in surface and sub-surface flow.

In the Nordic countries (Denmark, Norway, Finland and Sweden) the main reason for the significant loss of P from agricultural land is considered to be the high net input of P to the soil (calculated as 20 kg ha⁻¹ yr⁻¹) in recent decades [91]. Iserman [28] reports P surpluses of between 55, 71 and 88 kg ha⁻¹ yr⁻¹ for West Germany, East Germany and The Netherlands, respectively. For agricultural soils in the Netherlands,

Table 1. Percentages of soils in England and Wales with given Olsen extractable P contents (data from Skinner *et al.* [88])

Management	Percent soils with extractable P range (mg kg ⁻¹)					
	< 9	10-15	16-25	26-45	46-70	> 71
	%					
<i>Arable</i>						
1969-73	7	14	24	30	16	10
1974-78	4	13	26	36	14	7
1979-83	3	11	28	35	16	7
1984-88	3	11	31	34	16	6
<i>Ley-arable</i>						
1969-73	10	18	30	30	8	4
1974-78	9	22	31	26	7	6
1979-83	10	21	30	28	8	3
1984-88	10	18	34	28	8	3
<i>Grassland</i>						
1969-73	16	27	25	19	7	5
1974-78	15	22	30	24	8	2
1979-83	16	25	27	22	7	3
1984-88	17	23	30	22	6	2

Breeuwsma and Silva [8] estimated that in 1990 about 43% of those in grass and 82% of those in maize were P-saturated due to over-fertilization. In 1970, only 18% of these soils were P-saturated. Dutch soils are considered saturated when more than 25% of its sorption capacity is used [8]. The potential for P loss from these saturated soils is exacerbated by high water tables.

In the U.S.A., fertilizer P use declined from 2.2 to 1.8 million tonnes yr⁻¹ during 1978 to 1988 [6]. This decrease reflects efforts to reduce unnecessary applications and farmers' response to high soil P levels, policy changes, regulating price support, P fertilizer cost, removal of fertilizer subsidies, and production control measures. Even so, a large percentage of soils in areas with intensive cropping and livestock systems still have high or excessive soil test P, in terms of crop requirements [86]. However, in the U.S., high soil test P levels are a regional problem, with the majority of soils in the Great Plains for example, still requiring fertilizer P for optimum crop yields. Unfortunately, problems associated with high soil P are aggravated by the fact that many of these soils are located near P-sensitive water bodies, such as the Great Lakes, Chesapeake and Delaware Bays, Lake Okeechobee and the Everglades.

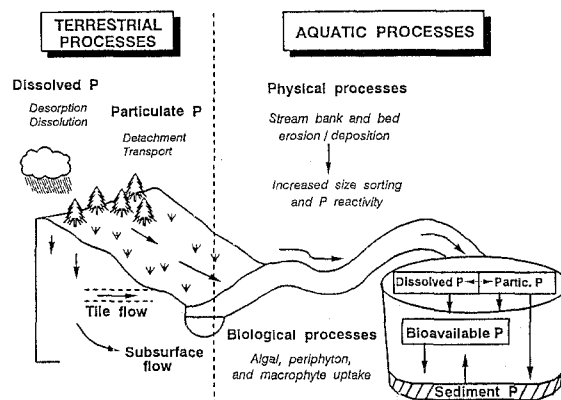


Fig. 1. Processes involved in the transport and bioavailability of P from agricultural land [adapted from 80].

Phosphorus transport in agricultural runoff

Transport processes

Phosphorus is transported in dissolved (DP) and particulate (PP) forms (Fig. 1; adapted from [80]). Dissolved P is mainly orthophosphate released from soil, vegetation, and applied fertilizer and manure and is

available for uptake by aquatic biota [62]. Particulate P is comprised of P sorbed by soil material, mineral P, and organic matter eroded during runoff, which can provide a long-term source of P to aquatic biota [83].

In contrast to runoff, sorption of P by P-deficient subsoils generally results in lower concentrations of DP in subsurface flow [5, 9, 68, 71, 79]. However, in organic or peaty soils, organic C may accelerate the downward movement of P together with organic acids and/or Fe and Al [14, 17, 26, 48, 87]. Similarly, P is more susceptible to movement through sandy soils with low P sorption capacities [2, 60, 71, 90] and in soils which have become waterlogged, where a decrease in Fe (III) is reduced to Fe (II) [19, 32, 63]. The migration of particulate P through fissured soils in summer and early autumn, in association with the dispersion of clay particles by percolating water, can also occur [29].

Once in stream flow, transformations between DP and PP can occur depending on relative DP and PP concentrations [91] (Fig. 1). These transformations are accentuated by the selective transport of fine material, which has a greater capacity to sorb or desorb P and will thus, be important in determining the bioavailability of P transported [74]. In addition, DP may be removed by stream macrophytes [24, 43, 99] and PP deposited or eroded from the stream channel with a change in flow velocity. Thus, changes in DP, PP, and resultant P bioavailability during channel flow, must be considered in assessing the impact of P loss from agricultural land on the trophic state of receiving water bodies.

Amounts transported

Runoff from uncultivated and agriculturally unimproved or pristine land is considered the background loading, which cannot be reduced. This source determines the natural trophic status of a lake or river. As we try to assess the impact of agricultural management on P loss in runoff, it becomes evident that little quantitative information is available on background losses of P from a given location prior to cultivation. Consequently, it is still difficult to quantify any increase in P loss following cultivation. These problems result mainly from the expensive and labor intensive nature of water quality monitoring studies, which are site-specific and impossible to replicate, due to spatial and temporal variations in climatic, edaphic, and agronomic conditions. Despite these problems, an investigation of published studies enables generalizations regarding

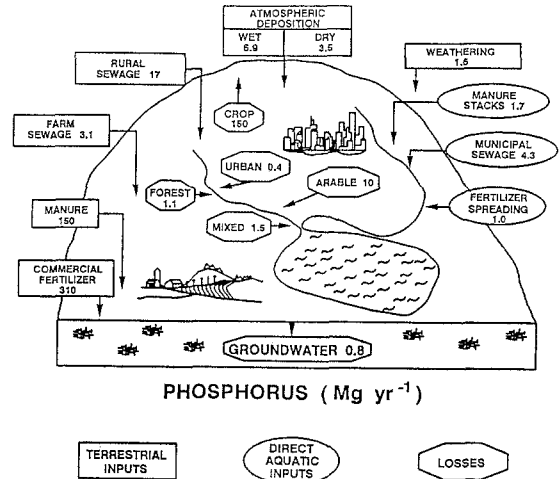


Fig. 2. Annual P inputs and losses from the drainage basin of Lake Ringsjon, Sweden [adapted from 70].

P transfer within and between ecosystems to be made. The complex nature of P transfers within terrestrial and aquatic ecosystems is shown in Fig. 2, which gives average annual P fluxes in the drainage basin of Lake Ringsjon, Sweden, as reported by Ryding *et al.* [70]. This information summarizes measured and calculated data collected over the last decade (1980–1990) by several Swedish scientists.

The atmospheric input of P to agricultural land is generally small compared to fertilizer and manure inputs (Fig. 2). However, the direct input of rainfall P to surface waters may be sufficient to enhance algal growth in certain situations. For example, Elder [15] estimated that rainfall P may account for up to 50% of the P entering Lake Superior. Since 25–50% of the TP in rainfall is dissolved, it is directly available to organisms in the lake [12, 53]. As a result, Schindler and Nighswander [72] attributed most of the enrichment of Clear Lake, Ontario, to rainfall and similar observations have been made for several Wisconsin lakes by Lee [38]. Ryding and Rast [69] report that the amount of P deposited from the atmosphere was dependent on the land-use of the surrounding area, emphasizing the often particulate nature of atmospheric P. Clearly, atmospheric inputs of P will be difficult to control, mainly because the source may be some distance from the impacted area. Also, political and regional boundaries can be crossed, and transfer pathways difficult to identify.

Increases in P loss in surface runoff have been measured after the application of fertilizer P (Table 2). Fertilizer P losses are influenced by the rate, time,

Table 2. Phosphorus loss from selected watersheds in the U.K. and the proportion attributable to agriculture

Watershed	Area (km ²)	Land use	P loss (kg ha ⁻¹) yr ⁻¹	Estimated agricultural contribution	Reference
				– % –	
Lough Neagh (1971–1979)	4453	74% grassland 19% rough grazing 7% arable Urban inputs	0.84 TP (mean) 0.57 DP	– 40	Foy <i>et al.</i> [18]
R. Main, Co Antrim	709	10% arable 52% grassland 24% rough grazing 10% woodland 4% urban	1.08 TP 0.65 DP	55 38	Smith [89]
Loch Leven	145	65% arable Urban and industrial input	0.65 TP 0.25 DP	51 34	Bailey-Watts and Kirika [4]
R. Wye – Frome	144	58% grassland 35% arable 5% forestry 1% rough grazing Urban inputs	0.47 DP	51	Houston and Brooker [25]
R. Wye – Trothy	142	81% grassland 14% arable 3% woodland 2% rough grazing Urban inputs	0.22 DP	86	Houston and Brooker [25]
Loch Lowes	14.9	3% arable 7% grassland 10% rough grazing 80% forest and heath No urban	0.18 DP	26	Harper and Stewart [21]
Loch Balgavies	24	75% arable 12% grassland 1% rough grazing 12% forest and heath No urban	0.27 DP	93	Harper and Stewart [21]
Loch Forfar	15.4	60% arable 14% grassland 1% forest and heath 25% urban	8.9 DP	4	Harper and Stewart [21]

and method of fertilizer application; form of fertilizer; amount and time of rainfall after application; and vegetative cover. The proportion of fertilizer P transported in runoff from undrained soils for the studies reported in Table 2, was generally greater from conventional compared to conservation tilled (no or zero till) watersheds. A greater proportion of fertilizer P was transported as PP than DP from the conventional

compared to no till and grassed watersheds (Table 2). However, fertilizer P application to no till corn reduced PP transport [46], probably due to an increased vegetative cover afforded by fertilization. Although it is difficult to distinguish between losses of fertilizer and native soil P without the use of expensive and hazardous radiotracers, the loss of fertilizer P in surface runoff is generally less than 5% of that applied.

Table 3. Effect of fertilizer P application on the loss of P in surface runoff

Land use	P applied	Concentration		Amount		Fertilizer loss		Reference
		Soluble	Partic.	Soluble	Partic.	Soluble	Partic.	
	kg ha ⁻¹ yr ⁻¹	mgL ⁻¹		kg ha ⁻¹ yr ⁻¹		%		
Contour corn	40	0.19	0.71	0.12	0.45			Burwell <i>et al.</i> [9], Minnesota
	66	0.25	1.27	0.15	0.76	0.1	1.2	
Grass	0	0.01	0.06	0.01	0.20			McColl <i>et al.</i> [44], New Zealand
	75	0.03	0.14	0.04	0.29	0.04	0.1	
No till corn silage	0	0.23	0.43	0.70	1.30			McDowell and McGregor [46], Mississippi
	30	0.39	0.49	0.80	1.00	0.3	+23.1 ^a	
No till corn grain	0	0.23	0.46	1.10	2.20			McDowell and McGregor [46], Mississippi
	30	0.57	0.51	1.80	1.60	2.3	+27.3 ^a	
Wheat – summer fallow	0	0.30	1.80	0.20	1.40			Nicholaichuk and Read [56], Western Canada
	54	3.70	7.40	1.20	2.90	1.9	2.8	
Grass	0	0.18	0.24	0.50	0.67			Sharpley and Syers [78], New Zealand
	50	0.98	0.96	2.80	2.74	4.6	4.1	

^aPercent decrease in P loss from fertilizer compared to check treatment.

The loss of P from several mixed land-use watersheds in the British Isles is generally less than 1 kg P ha⁻¹ yr⁻¹ (Table 3). The estimated contribution of nonpoint sources from agriculture to these losses is variable (5 to 90%) but can be the major contributor (Table 3).

In small watershed studies, greater P concentrations have been measured in runoff from heavily stocked watersheds compared to lightly stocked watersheds [103]. The amount of P lost in surface runoff from land receiving surface application of livestock manures has been shown to depend not only on the rate and timing of the manure application but also on the time interval between application and the runoff event [84]. Highest P losses, representing up to nearly 20% of the P applied, have occurred on sloping, poorly drained and/or frozen soils [33, 34, 95]. Phosphorus concentrations as high as 30 mg L⁻¹ were recorded soon after pig and cattle slurry applications to grassland in Ireland and were greater than 1 mg L⁻¹ even six weeks after application [85].

The loss of P in subsurface runoff is appreciably lower than that in surface runoff because of P sorption from infiltrating water as it moves through the soil profile (Table 4). Subsurface runoff is discussed as accelerated and natural subsurface runoff, where accelerated flow is percolating water intercepted by

artificial drainage systems, such as tile or mole drain, which accelerate its movement into streams. In general, P concentrations and losses in natural subsurface runoff were lower than in tile drainage (Table 4). A longer time of contact between the subsoil material and natural subsurface runoff compared to tile drainage will result in a greater retention of P by soil and lower DP in natural subsurface runoff. Increased sorption of P from percolating water also accounted for lower P loads from 1.0 m deep (0.32 kg ha⁻¹ yr⁻¹) than from 0.6 m deep (0.83 kg ha⁻¹ yr⁻¹) tiles draining a Brookston clay soil under alfalfa in Ontario, Canada [11]. For the shallower drains, TP loads were about 1% of fertilizer P applied (30 kg P ha⁻¹ yr⁻¹) whereas 1 m deep tiles exported about 0.6% of that applied.

Significant increases in P concentrations in drainage water (up to 10 mg L⁻¹) have also been observed where diluted livestock slurries have been applied to cracking clay soils or to land which is intensively underdrained with permeable backfill [42]. Also, Sharpley and Syers [79] found that, in 4 weeks following grazing, DP loss in drainage from 45-cm deep tiles (68 g ha⁻¹) was 50% greater than from ungrazed plots (45 g ha⁻¹). The response of DP in tile drainage to grazing was rapid with maximum concentrations (0.25 mg L⁻¹) occurring only 1 week after grazing [79].

Table 4. Effect of fertilizer P application on the loss of P in subsurface runoff

Land use	P applied	Concentration		Amount		Fertilizer loss		Reference
		Soluble	Partic.	Soluble	Partic.	Soluble	Partic.	
	kg ha ⁻¹ yr ⁻¹	mgL ⁻¹		kg ha ⁻¹ yr ⁻¹		%		
Alfalfa	0	0.180	–	0.12	0			Bolton <i>et al.</i> [7],
(tile drainage)	29	0.210	–	0.19	–	1.0	–	Canada
Continuous corn	40	0.007	–	0.03	–	–	–	Burwell <i>et al.</i> [9],
	66	0.009	–	0.04	–	–	–	Iowa
Terraced corn	67	0.028	–	0.17	–	–	–	Burwell <i>et al.</i> [9], Iowa
Bromegrass	40	0.005	–	0.03	–	–	–	Burwell <i>et al.</i> [9], Iowa
Continuous corn	0	0.20	0.100	0.13	0.29			Culley <i>et al.</i> [11],
(tile drainage)	30	0.110	0.360	0.20	0.42	0.2	0.4	Canada
Blue grass sod	0	0.02	0.15	0.06	0.09	–	–	Culley <i>et al.</i> [11],
(tile drainage)	30	1.01	3.29	0.16	0.21	0.3	0.4	Canada
Oats	0	0.02	0.09	0.10	0.19			Culley <i>et al.</i> [11],
(tile drainage)	30	0.42	1.10	0.20	0.30	0.3	0.4	Canada
Alfalfa	0	0.02	0.011	0.012	0.020			Culley <i>et al.</i> [11],
(tile drainage)	30	0.37	1.03	0.20	0.31	0.3	0.3	Canada
Corn	17	0.018	0.043	0.005	0.02	–	–	Hanway and Lafen [20],
(tile drainage)	42	0.000	0.000	0.000	0.00	–	–	Iowa
	44	0.004	0.024	0.004	0.04	–	–	
Grass	0	0.020	0.022	0.04	0.44			Sharpley and Syers [78],
	50	0.033	0.019	0.12	0.07	0.2	+7.4 ^a	New Zealand
Grass	0	0.064	0.072	0.08	0.09			Sharpley and Syers [78],
(tile drainage)	50	0.190	0.161	0.44	0.37	0.7	0.6	New Zealand

^aPercent decrease in P loss from fertilizer compared to check treatment.

Clearly, the main factors controlling these nonpoint P losses from agricultural land include the relative importance of surface and subsurface runoff in a watershed; fertilizer and manure inputs of P; and runoff and erosion potential as influenced by land management. These losses are often small (generally < 2 kg P ha⁻¹) and represent a minor proportion of fertilizer or manure P applied (generally < 5%). Thus, these losses are not of economic importance to farmers in terms of irreplaceable fertility. However, they may contribute to the P related eutrophication of surface waters and are, thus, of environmental and off-site economic importance. In as much, environmental impacts of agricultural P loss raise several concerns that must be addressed.

Current concerns

The main concerns currently facing the environmentally sound management of agricultural P involve (1) the accumulation of soil P in excess of crop require-

ments, which increases the potential for P loss in runoff or drainage water, and (2) land management practices which affect the ease with which P enters water courses. In most cases, the accumulation of soil P has occurred in areas of intensive crop and livestock production [8, 16, 30]. Thus, current environmental P concerns revolve around soil P fertility, animal manure, and land management.

Fertility management

Continual long-term application of fertilizer and manures at levels exceeding crop requirements can raise soil test P above those levels required for economically optimum crop yields in the runoff-sensitive portion of surface soil (0 to 2 cm). The build-up of P in soil is accentuated by the poor efficiency with which P, added as fertilizer or manure, is utilized by agricultural crops. This build-up can be rapid where there is an economic gain to large applications of fresh P fertilizer (eg., to potatoes) (Table 5). Where animal manures

Table 5. Potential input and removal of P by different crops receiving recommended rates of fertilizer P in the United Kingdom (from Ministry of Agriculture, Fisheries, and Food [50])

Crop	Potential input	Potential removal ^a	Balance
		kg P ha ⁻¹ yr ⁻¹	
Winter wheat	28	27 (7)	+1
Spring barley	19	19 (5)	0
Potatoes	91	17 (40)	+74
Sugar beet	30	14 (40)	+16
Oilseed rape	27	21 (3)	+6
Peas	26	14 (3.5)	+12
Beans	27	14 (3)	+13
Cabbage	39	20 (50)	+19

^a Number in parenthesis is average crop yield in t ha⁻¹.

are also applied, potential accumulation of P is even greater (Table 6). Clearly, P inputs of up to 190 kg ha⁻¹ yr⁻¹ greater than crop removal can quickly create environmental problems. However, considerable time is required for significant depletion of excessive soil test P. Johnston [30] and McCollum [45] found half-lives of about 9 years for Olsen P contents of a clay loam soil and Mehlich III contents of a sandy soil, respectively.

Once soil test P levels become excessive, the potential for P loss, if runoff and erosion occur, is greater than any agronomic benefits of further P applications. This is due to the dependence of P loss in runoff on surface soil P content. A highly significant linear relationship was obtained between the soil test P content (Bray I) of surface soil (5 cm) and the DP concentration of runoff from drained and undrained field plots in New Zealand (Fig. 3; from [82]). The consistently higher DP concentration in surface runoff from undrained than drained fields reflects the desorption of P from the higher sediment loads found in these waters. Several other studies have also reported a close dependence of the DP concentration of runoff on soil test P [58, 66, 73].

Because of the variable path and time of water flow through a soil with subsurface drainage, factors controlling DP in subsurface waters are more complex than for surface runoff. However, the DP concentration of tile flow in New Zealand was related to the Bray-1 P content of soil at the tile drain depth (40 to 50 cm) (Fig. 3; from [82]). A similar dependence of DP concentration in tile drainage on the P sorption-desorption properties of subsoil material was found for Histosols

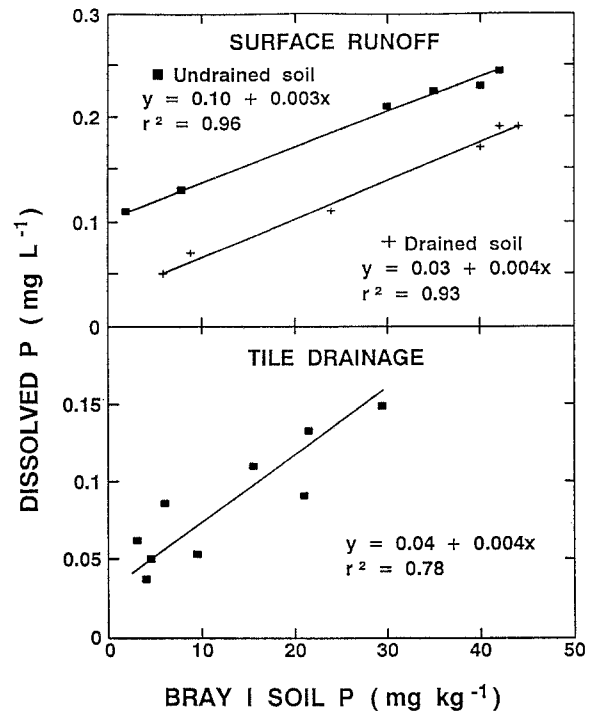


Fig. 3. Relationship between the Bray-1 P content of surface (0–5 cm) and subsurface (40–50 cm) soil and the dissolved P concentration of surface runoff and tile drainage, respectively, from several unfertilized and fertilized soils in New Zealand [adapted from 79, 82].

in Florida [26], New York [10, 14], and Ontario [57], and for Haploquolls in Ontario and Michigan [11].

Manure management

Excessive soil P has occurred to a greater extent in livestock systems, which on economic grounds need to operate on less land area. Thus, development of manure management plans that are environmentally sound is a concern facing an increasing number of landowners.

Animal manure can be a valuable resource integrated in cost-effective Best Management Practices (BMPs). In many areas, manure applications have improved soil structure and increased vegetative cover, thereby reducing runoff and erosion potential. However, in areas where a large number of confined animal operations are located, the amount of nutrients in manure often exceeds local crop requirements and the area of land available for application. Clearly, economics is the crucial issue facing efficient utilization of manure. This involves applying manure on a P

Table 6. Potential input of P in fertilized and animal manures in areas of intensive livestock production in Italy (Po Basin) and the Netherlands (Sand districts) and removal in maize and wheat [8]

Region	Potential input		Potential removal ^a		Balance	
	Fertilizers	Manures	Maize	Wheat	Maize	Wheat
kg P ha ⁻¹ yr ⁻¹						
<i>Italy</i>						
Piemonte	22	16	40	30	-2	+8
Lombardia	31	30	40	30	+21	+31
Veneto	35	26	40	30	+21	+31
Emilia-Romagna	27	19	40	30	+6	+16
<i>Netherlands</i>						
Salland-Twente	15	85	40	30	+60	+70
West Veluwe	15	174	40	30	+149	+159
Meijerij	15	130	40	30	+105	+115
South Peel	15	204	40	30	+179	+189

^aMaize and wheat yields of 10 and 5 t ha⁻¹ yr⁻¹, respectively.

rather than N basis, limiting the area of land available for application, transporting manure to non-producing areas, and linking the number of animals to the area of land available for application of manure. Thus, we can develop BMPs for manure but should we expect farmers to bear the total cost of these programs?

In Europe and the U.S., rates for application of animal manure have been based on the N needs of the crop in order to minimize nitrate losses by leaching and the potential for ground water contamination. In most cases, this strategy has led to an increase in soil P levels in excess of crop requirements due to the generally lower ratio of N:P added in manure (4:1) than taken up by crops (8:1).

Site hydrology will also be important in determining manure application strategies. If the potential for nitrate leaching from an application site exists, N should be a priority management consideration. However, if runoff and erosion potential exceeds leaching potential then P should be the main element driving application rates, as this strategy will mitigate the excessive build up of soil P and at the same time lower the risk for nitrate leaching to ground water. A soil P-based strategy could eliminate much of the land area with a history of continual manure application from further additions, since many years are required to lower soil P levels once they become excessive. This would force farmers to identify larger areas of land on which to utilize the generated manure, further exac-

erbatng the problem of local land area limitations. In addition, farmers relying on manure to supply most of their crop N requirements may be forced to buy fertilizer N to supplement foregone manure N. Using a P-based strategy may resolve potential environmental issues but could place additional economic burdens on farmers.

Options for efficient utilization of manure include basing application rates on site susceptibility to runoff (P-based strategy) or leaching (N-based strategy); linking the number of animals to area of land available for manure utilization; formation of co-operatives that can more economically compost or concentrate manure to increase the economic distance it can be transported from its source, thus increasing the area of land for application; establishment of cost-sharing programs so that both the consumer and producer share the economic burden of environmental sustainability; and expansion of education and extension programs highlighting the nutritive and mulching value of manure to non-producing farmers.

Land management

Although land management encompasses both fertility and manure management, additional broader land management issues must be considered. For example, conservation tillage may increase crop yields by enhancing soil moisture, while in some cases weeds

may reduce crop yields and quality [83]. Further, conservation tillage may reduce erosion and TP loss in runoff compared to conventional tillage but increase the potential for DP loss and nitrate leaching [78, 83, 93]. Other land management changes such as set-aside land, residue burning, wetland classification, animal manure incorporation, and weed management will also impact environmental issues.

Clearly, changes in land management must be considered not only in terms of whether they improve productivity but if they positively or negatively influence environmental sustainability. As a result, land management policy decisions will be required to address the potential for severe water quality problems. This will be of particular relevance to management systems involving manural inputs of P. Fertilizer inputs can be easily adjusted to accurately balance crop uptake of P, however, manure often provides excess P. For example, confined animal operations should develop a comprehensive waste management plan tailored to local land limitations, involving several options discussed in the previous section. Thus, land management planning will have to address the potential for water quality issues as well as maximizing profitability. However, if the cost of the environmental effects of P loss in runoff are included in land management decisions, this may lower economically optimum crop yields compared to traditional yields. Even so, it is possible that land management will only change if there are clear economic (or legislative) incentives for them to do so. We hope targeting flexible, cost-effective BMPs will minimize P loss in runoff and lead to environmentally sustainable land management through recommendation rather than regulation.

Future challenges

To achieve environmentally-sound P management in agriculture, we must be able to identify soil P levels that are of environmental concern; assess sites and management practices within a watershed that are vulnerable to P loss; and target remedial measures to minimize P loss in runoff and erosion. However, in doing this we must ensure the economic viability of farming.

Environmental tests for soil phosphorus

Soil P tests have provided farmers with an indication of how much plant available P is present in a soil and consequently how much fertilizer to apply to obtain

the desired crop yields. However, as we move from agronomic to environmental concerns with soils containing P levels in excess of crop requirements, will current soil test P methods assessing plant availability, estimate P forms that can accelerate eutrophication? If not, are appropriate methods available? Environmental soil tests for P must estimate the bioavailability of soil, sediment, or runoff P to aquatic organisms and the long-term capacity of a soil to retain P against leaching.

The amount of P in soil, sediment, or runoff that is potentially available for algal uptake can be quantified by algal assays, but these require up to 100-d incubations [47]. Thus, more rapid chemical extractions, such as NaOH [13], NH_4F [64], and ion exchange resin [27] have been developed. More recently, Sharp-ley [76] showed that the amount of P removed from runoff sediment by iron oxide-impregnated filter paper (Fe-oxide strip) was related ($r^2 = 0.92\text{--}0.95$) to the growth of several algal species incubated with runoff as the sole source of P. As the strips act as a P-sink, they simulate P removal from soil or sediment-water samples by plant roots and algae. Thus, use of the Fe-oxide strip method has a stronger theoretical justification to estimate bioavailable P than do chemical extractants. The method may have potential use as an environmental soil P test to identify soils liable to enrich runoff with sufficient P to accelerate eutrophication.

In addition to bioavailable P, environmental soil tests will need to estimate the long-term capacity of a soil to retain P against leaching. For example, estimates of the P-loading capacity of soils receiving continual applications of P in manure or waste water will aid development of sustainable management systems. This capacity is commonly estimated by sorption isotherms that can be used to derive sorption maxima for soil horizons. These isotherms require equilibration of soil with a solution of P for up to 6 d, and are, thus, not appropriate for routine soil testing. However, Bache and Williams [3] suggested that a single-point isotherm can provide a reasonably accurate estimate of soil P sorption maxima.

The additional analytical workload for a soil test laboratory to conduct environmental tests for P, along with a lack of rigorous field testing to determine critical P levels associated with eutrophication, will limit their widespread adoption in the near future [85]. However, several studies have shown the amounts of P extracted by standard soil tests (Bray, Mehlich, and Olsen) are related to bioavailable P estimated by either Fe-oxide strip or NaOH methods [75, 105]. Also, P sorption parameters can be approximated by simpler

water extractions [52, 81]. Consequently, in areas with P-related water quality problems, soil test laboratories could use routine soil tests as surrogates to provide preliminary rankings of the algal available P content and identify those on which environmental tests should be conducted.

Assessing site vulnerability to phosphorus loss

Soil testing alone cannot assess the potential for soil P from an individual site or watershed to play a significant role in surface water eutrophication. Any environmental soil P test must be linked to site assessment of drainage, runoff and erosion potential and management factors affecting the vulnerability for P transport from a site. Strategies to minimize P loss in runoff will be most effective if they are targeted to sensitive or source areas identified within a watershed rather than through widespread implementation over broad areas.

Lemunyon and Gilbert [39] developed a P indexing system to identify sites vulnerable to P loss in runoff. The index rates source (soil test P and fertilizer and manure management) and transport factors (runoff and erosion potential) of a site to provide a numerical value, ranking site vulnerability to P loss in runoff [84]. The index is intended for use as a tool for field personnel to easily identify agricultural areas or practices that have the greatest potential to accelerate eutrophication [84]. It is hoped that this will identify management options available to land users that will allow them flexibility in developing remedial measures.

Minimizing phosphorus losses

Minimizing the loss of agricultural P in runoff can be accomplished by P-source management and erosion and runoff control. Source management of P includes basing application rates on eutrophic rather than agronomic soil test P guidelines and the possible use of slow-release P fertilizers. In addition, subsurface placement of P away from the zone of removal in runoff will reduce the potential for P loss. Where practical and economically viable, subsurface placement may also increase crop yield response to P additions.

The most effective measure to reduce total P losses in runoff involves the implementation of conservation or reduced tillage, although DP losses can be increased compared to conventional cultivation [77, 83]. The surface soil accumulation of added P and crop residues can be a source of P to runoff that would otherwise be reduced by tillage incorporation [83]. Thus, it may

be necessary to incorporate surface accumulations of P in the soil profile by occasional plowing. Conservation tillage has also been shown to enhance nitrate movement to ground water compared to conventional tillage [77, 93]. Clearly, the potential for enhanced DP loss and nitrate leaching should be accounted for in assessing the effectiveness and potential water quality benefits (reduced soil erosion and total P loss) of conservation tillage measures.

The loss of P by erosion and runoff may also be reduced by terracing, contour tillage, cover crops, grassing of valley floors, and creation of targeted riparian zones [51, 54]. These practices efficiently reduce PP losses, and thereby the long-term potential for eutrophication, due to recycling of deposited PP. However, these practices are less efficient at reducing DP loss and in many cases the bioavailability of P transported is increased [83]. Often measures to reduce P inputs to aquatic systems have had less effect on the degree of eutrophication than expected, due to an increased bioavailability of P entering the system as well as internal recycling of P. Thus, the environmentally-sound management of agricultural P will involve both P-source management as well as erosion and runoff control. Clearly, a combination of remedial measures will be necessary to reduce eutrophication to acceptable levels.

Research priorities

The benefits of remedial measures on freshwater eutrophication may not be immediately visible. Thus, further research should emphasize the long-term economic and environmental benefits of these measures. Further development of procedures to estimate the potential of soils to enrich the P content of runoff or retain P in the profile during drainage is needed. Even if current soil test procedures are used as surrogates for environmental tests until improved methods are available, field calibration is essential. In other words, site specific threshold soil P levels are needed above which the potential for eutrophication exceeds any agronomic benefit of P application; and above which it is recommended that either no P be applied or that the rate be less than crop removal rates.

Further development of P indexing procedures to target source areas within watersheds must address the ability of P losses to cause eutrophication. This will involve evaluation of the proximity of a P source to a water course or body, biosensitivity of a water body to

P inputs, and the major use of affected water bodies and desired trophic status.

In many cases, P-related eutrophication is accelerated by the land application of excessive amounts of animal manure in localized areas. Consequently, future research must address the technical and economic feasibility of reducing P solubility in manures and treated soils, reducing transportation costs, and developing alternative uses for manure. Alternatives include power generation, livestock feed, and market garden or home use of composted material.

Finally, in the past we have tended to assess the effectiveness of individual practices to reduce P loss from agricultural land. Future field research should evaluate remedial systems which incorporate several measures to minimize soil P accumulations, runoff and erosion potential, and delivery of P from its source to water body. Land use and management practices are constantly changing in Europe and the U.S.A. For example, conservation tillage, set-aside land, land-burning bans, and irrigation are all new developments which will impinge on P transport from agricultural land. Thus, the effect of these changes on the development of environmentally-sound management systems for agricultural P should be considered.

References

- Aarts HFJ, Biewinga EE and Van Keulen H (1993) Dairy farming systems based on efficient nutrient management. *Netherlands J of Agric Sci* 40: 285–299.
- Adriano DC, Novak LT, Erikson AE, Wolcott AR and Ellis BG (1975) Effect of long-term disposal by spray irrigation of food processing wastes on some chemical properties of the soil and subsurface water. *J Environ Qual* 4: 242–248
- Bache BW and Williams EG (1971) A phosphate sorption index for soils. *J Soil Sci* 22: 289–301
- Bailey-Watts AE and Kirika A (1987) A reassessment of phosphorus inputs to Loch Leven (Kinross, Scotland): Rationale and an overview of results on instantaneous loadings with special reference to runoff. *Trans Royal Soc of Edinburgh. Earth Sci* 78: 351–367
- Baker JL, Campbell KL, Johnson HP and Hanway JJ (1975) Nitrate, phosphorus and sulfate in subsurface drainage waters. *J Environ Qual* 4: 406–412
- Berry JT and Hargett NL (1989) Fertilizer summary data, 1988 TVA/NFDC-89/3, Bulletin Y-209. Natl Fert Develop Center, TVA, Muscle Shoals, AL, pp130
- Bolton EF, Aylesworth JW and Hove FR (1970) Nutrient losses through tile drainage under three cropping systems and two fertility levels on a Brookston clay soil. *Can J Soil Sci* 50: 272–279
- Breeuwsma A and Silva S (1992) Phosphorus fertilization and environmental effects in The Netherlands and the Po region (Italy). Rep. 57. Agric. Res. Dep. The Winand Staring Centre for Integrated Land, Soil and Water Research. Wageningen, The Netherlands
- Burwell RE, Schuman GE, Heinemann HG and Spomer RG (1977) Nitrogen and phosphorus movement from agricultural watersheds. *J Soil and Water Conserv* 32: 226–230
- Cogger G and Duxbury JM (1984) Factors affecting phosphorus losses from cultivated organic soils. *J Environ Qual* 13: 111–114
- Culley JLB, Bolton EF and Bernyk V (1983) Suspended solids and phosphorus loads from a clay soil: I. Plot studies. *J Environ Qual* 12: 493–498
- Delumyea RG and Petel RL (1977) Atmospheric inputs of phosphorus to southern Lake Huron, April–October, 1975. U.S. EPA Report No. 600/3–77–038. Duluth, MN
- Dorich RA, Nelson DW and Sommers LE (1980) Algal availability of sediment phosphorus in drainage water of the Black Creek watershed. *J Environ Qual* 9: 557–563
- Duxbury JM and Peverly JH (1978) Nitrogen and phosphorus losses from organic soils. *J Environ Qual* 7: 566–570
- Elder FC (1975) International Joint Commission Program for Atmospheric Loading of the Upper Great Lakes. Second Interagency Committee on Marine Science and Engineering Conference on the Great Lakes, Argonne, IL
- Federico AC, Dickson KG, Kratzer CR and Davis FE (1981) Lake Okeechobee water quality studies and eutrophication assessment. Tech. Pub. 81–2. South Florida Water Management District, West Palm Beach, FL. 270 pp
- Fox RL and Kamprath EJ (1971) Adsorption and leaching of P in acid organic soils and high organic matter sand. *Soil Sci Soc Am Proc* 35: 154–156
- Foy RH, Smith RV, Stevens RJ and Stewart DA (1982) Identification of factors affecting nitrogen and phosphorus loadings to Lough Neagh. *J Environ Manag* 15: 109–129
- Gotoh S and Patrick WH Jr (1974) Transformations of iron in a waterlogged soil as influenced by redox potential and pH. *Soil Sci Soc Am Proc* 38: 66–71
- Hanway, JJ and Lafen JM (1974) Plant nutrient losses from tile outlet terraces. *J Environ Qual* 7: 208–212
- Harper DM and Stewart DM (1987) The effects of land use upon water chemistry, particularly nutrient enrichment, in shallow lowland lakes: comparative studies of three lochs in Scotland. *Hydrobiologica* 148: 211–229
- Haynes RJ and Williams PH (1992) Long term effect of superphosphate on accumulation of soil phosphorus and exchangeable actions on a grazed, irrigated pasture site. *Plant and Soil* 42: 123–133
- Hillbricht-Ilkowska A (1988) Transport and transformation of phosphorus compounds in watersheds of Baltic Lakes. p. 193–206. In Tiessen H (ed) *Phosphorus Cycles in Terrestrial and Aquatic Ecosystems. Regional Workshop 1: Europe. May 1988, Czerniejewo, Poland. Published by Saskatchewan Institute of Pedology, Saskatoon, Canada*
- House WA and Casey H (1989) Transport of phosphorus in rivers. pp. 253–282. In Tiessen H (ed) *Phosphorus Cycles in Terrestrial and Aquatic Ecosystems. Proceedings of the SCOPE and UNEP 7: Europe. May 1 – May 6, 1988, Czerniejewo, Poland*
- Houston JA and Brooker MP (1981) A comparison of nutrient sources and behaviour in two lowland sub-catchments of the River Wye. *Water Res* 15: 49–57
- Hortensteine CC and Forbes RB (1972) Concentration of nitrogen, phosphorus, potassium and total soluble salts in soil solution samples from fertilized and unfertilized histosols. *J Environ Qual* 1: 466–449

27. Huettl PJ, Wendt RC and Corey RB (1979) Prediction of algal available phosphorus in runoff suspension. *J Environ Qual* 4: 541–548
28. Iserman K (1990) Share of agriculture in nitrogen and phosphorus emissions into the surface waters of Western Europe against the background of their eutrophication. *Fert Res* 26: 253–269
29. Johnes PJ (1990) An investigation of the effects of land use upon water quality in the Windrush catchment. D. Phil Thesis, University of Oxford
30. Johnston AE (1989) Phosphorus cycling in intensive arable agriculture. p. 123–126. In: Tiessen H (ed) Proceedings of a SCOPE and UNEP Workshop, May 1 to May 6, 1988, Czerniejewo, Poland
31. Kauppi L (1990) Hydrology: Water quality changes pp. 43–66. In Solomon AM and Kauppi L (eds) Towards Ecological Sustainability in Europe. International Institute for Applied System Analysis, Laxemburg, Austria
32. Khalid RA, Patrick WH Jr and Delaune RD (1977) Phosphorus sorption characteristics of flooded soils. *Soil Sci Soc Am J* 41: 301–305
33. Kiely PV (1981) Catchment pollution. p. 87–91. In Logan JC (ed) Nitrogen Losses and Surface Runoff from Landspreading of Manures. Martinus Nijhoff, The Hague
34. Klausner SD Zaverman PJ and Ellis DF (1976) Nitrogen and phosphorus losses from winter disposal of dairy manure. *J Environ Qual* 5: 46–49
35. Kotak BG, Kenefick SL, Fritz DL, Rousseaux CG, Prepas EE and Hrudey SE (1993) Occurrence and toxicological evaluation of cyanobacterial toxins in Alberta lakes and farm dugouts. *Water Res* 27: 495–506
36. Krogstad T and Lovstad O (1989) Erosion, phosphorus, and phytoplankton response in rivers of South-eastern Norway. *Hydrobiologia* 183: 33–41
37. Lawton LA and Codd GA (1991) Cyanobacterial (blue-green algal) toxins and their significance in U.K. and European waters. *J Inst Wat Environ Manag* 5: 460–465
38. Lee GF (1973) Role of phosphorus in eutrophication and diffuse source control. *Water Res* 7: 111–128
39. Lemunyon JL and Gilbert RG (1993) Concept and need for a phosphorus assessment tool. *J Prod Agric*. 6: 483–486
40. Levine SL and Schindler DW (1989) Phosphorus nitrogen and carbon dynamics of experimental lake 303 during recovery from eutrophication. *Can J Fish Aquat Sci* 46: 2–10
41. Lund JWG and Moss B (1990) Eutrophication in the United Kingdom, Trends in the 1980's. A report to the Soap and Detergent Industry, Hayes, Middlesex, United Kingdom. 81 pp
42. McAllister JSV (1976) Studies in Northern Ireland on problems related to the disposal of slurry. pp. 418–431. London, United Kingdom. Agriculture and Water Quality, Technical Bulletin 32, Ministry of Agriculture, Fisheries and Food, London, United Kingdom
43. McColl RHS (1974) Self-purification of small freshwater streams: Phosphate, nitrate and ammonia removal. *NZ J Mar Freshwater Res* 8: 375–388
44. McColl RHS, White E and Gibson AR (1977) Phosphorus and nitrate runoff in hill pasture and forest catchments, Taita, New Zealand. *NZ J Mar Freshwater Res* 11: 729–744
45. McCollum RE (1991) Buildup and decline in soil phosphorus: 30-year trends on a Typic Umprabuult. *Agron J* 83: 77–85
46. McDowell LL and McGregor KC (1984) Plant nutrient losses in runoff from conservation tillage corn. *Soil Tillage Res* 4: 79–91
47. Miller WE, Greene JC and Shiroyama T (1978) The *Selenastrum capricornutum* Printz algal assay bottle test and data interpretation protocol. U.S. EPA, Tech Rep EPA-600/9-78-018, Corvallis, OR
48. Miller MH (1979) Contribution of nitrogen and phosphorus to subsurface drainage water from intensively cropped mineral and organic soils in Ontario. *J Environ Qual* 8: 42–48
49. Ministry of Agriculture, Fisheries and Food (1985) Phosphate and Potash for Rotations. Booklet 2496, Ministry of Agriculture, Fisheries and Food, London, United Kingdom. 13 pp
50. Ministry of Agriculture, Fisheries and Food (1991) Code of Good Agricultural Practice for the Protection of Water. Ministry of Agriculture, Fisheries and Food, London, United Kingdom. 80pp
51. Morgan RPC (1992) Soil conservation options in the U.K. *Soil Use and Manag* 8: 176–180
52. Mozaffari M and Sims JT (1994) Phosphorus availability and sorption in an Atlantic Coastal Plain watershed dominated by animal-based agriculture. *Soil Sci*. 157: 97–107
53. Murphy TJ and Doskey PV (1975) Inputs of phosphorus from precipitation to Lake Michigan. U.S. EPA Report No 600/3-75-005. Duluth, MN
54. Muscutt AD, Harris GL, Bailey SW and Davies DB (1993) Buffer zones to improve water quality: a review of their potential use in U.K. agriculture. *Agric Ecosyst and Environ* 45: 59–77
55. National Rivers Authority (1990) Toxic Blue-green Algae. NRA Water Quality Series No. 2 Peterborough, United Kingdom. 125 pp
56. Nicholaichuk W and Read DWL (1978) Nutrient runoff from fertilized and unfertilized fields in western Canada. *J Environ Qual* 7: 542–544
57. Nicholls KH and MacCrimmon HR (1974) Nutrients in subsurface and runoff waters of the Holland Marsh, Ontario. *J Environ Qual* 3: 31–35
58. Olness AE, Smith SJ, Rhoades ED and Menzel RG (1975) Nutrient and sediment discharge from agricultural watersheds in Oklahoma. *J Environ Qual* 4: 331–336
59. Organization for Economic Cooperation and Development (1982) Eutrophication of Waters: Monitoring, Assessment, and Control. OECD Paris, France
60. Ozanne PG, Kirton DJ and Shaw TC (1961) The loss of phosphorus from sandy soils. *Aust J Agric Res* 12: 409–423
61. Palmstrom NS, Carlson RE and Cooke GD (1988) Potential links between eutrophication and formation of carcinogens in drinking water. *Lake Reserv Manag* 4: 1–15
62. Peters RH (1981) Phosphorus availability in Lake Memphremagog and its tributaries. *Limnol Oceanogr* 26: 1150–1161
63. Ponnampuruma FN (1972) The chemistry of submerged soils. *Soil Sci Soc Am Proc* 41: 305–310
64. Porcella DB, Kumazar JS and Middlebrooks EJ (1970) Biological effects on sediment-water nutrient interchange. *J Sanit Eng Div Proc Am Soc Civil Eng* 96: 911–926
65. Rast W and Lee GF (1978) Summary analysis of the North American (U.S. Portion) OECD eutrophication project: Nutrient loading – lake response relationships and trophic state indices. EPA 600/3-78-008, U.S. EPA, Corvallis, OR
66. Romkens MJM and Nelson DW (1974) Phosphorus relationships in runoff from fertilized soil. *J Environ Qual* 3: 10–13
67. Rohlich GA and O'Connor DJ (1980) Phosphorus management for the Great Lakes. Final Rep., Phosphorus Manage-

- ment Strategies Task Force. Int Joint Commission (IJC). PLU-ARG Tech. Rep
68. Ryden JC, Syers JK and Harris RF (1973) Phosphorus in runoff and streams. *Adv Agron* 25: 1–45
 69. Ryding SO and Rast W (1989) *The Control of Eutrophication of Lakes and Reservoirs. Man and the Biosphere Series, Volume 1. Unesco, Paris and the Parthenon Publishing Group, Paris, France*
 70. Ryding SO, Enell M and Wennberg L (1990) Swedish agricultural nonpoint source pollution: A summary of research and findings. *Lake and Reserv Manag* 6: 207–217
 71. Sawhney BL (1977) Predicting phosphate movement through soil columns. *J Environ Qual* 6: 86–89
 72. Schindler DW and Nighswander JE (1970) Nutrient supply and primary production in Clear Lake, eastern Ontario. *J Fish Res Board Can* 27: 260–262
 73. Schreiber JD (1988) Estimating soluble phosphorus (PO_4 -P) in agricultural runoff. *J Miss Acad Sci* 33: 1–15
 74. Sharpley AN (1985) The selective erosion of plant nutrients in runoff. *Soil Sci Soc Am J* 49: 1527–1534
 75. Sharpley AN (1991) Soil phosphorus extracted by iron-aluminum-oxide-impregnated filter paper. *Soil Sci Soc Am J* 55: 1038–1041
 76. Sharpley AN (1993) An innovative approach to estimate bioavailable phosphorus in agricultural runoff using iron oxide-impregnated paper. *J Environ Qual* 22: 597–601
 77. Sharpley AN and Smith SJ (1994) Wheat tillage and water quality in the Southern Plains. *Soil Tillage Res* 30: 33–48
 78. Sharpley AN and Syers JK (1979a) Phosphorus inputs into a stream draining an agricultural watershed: II. Amounts and relative significance of runoff types. *Water, Air and Soil Pollut* 11: 417–428
 79. Sharpley AN and Syers JK (1979b) Loss of nitrogen and phosphorus in tile drainage as influenced by urea application and grazing animals. *NZ J Agric Res* 22: 127–131
 80. Sharpley AN, Daniel TC and Edwards DR (1993) Phosphorus movement in the landscape. *J Prod Agric* 6: 492–500
 81. Sharpley AN, Reed LW and Simmons DK (1982) Relationships between available soil phosphorus forms and their role in water quality modeling. *Oklahoma Agric Expt Sta Tech Bull T-157. Oklahoma State Univ., Stillwater, OK.* 40 pp
 82. Sharpley AN, Tillman RW and Syers JK (1977) Use of laboratory extraction data to predict losses of dissolved inorganic phosphate in surface runoff and tile drainage. *J Environ Qual* 6: 33–35
 83. Sharpley AN, Smith SJ, Jones OR, Berg WA and Coleman GA (1992) The transport of bioavailable phosphorus in agricultural runoff. *J Environ Qual* 21: 30–35
 84. Sharpley AN, Chapra SC, Wedepohl R, Sims JT, Daniel TC and Reddy KR (1994) Managing agricultural phosphorus for protection of surface waters: Issues and options. *J Environ Qual* 23: 437–451
 85. Sherwood MT and Fanning A (1981) Nutrient content of surface runoff water from land treated with animal wastes. pp. 5–17. In Brogan JC (ed) *Nitrogen Losses and Surface Runoff from Land Spreading of Manures.* Martinus Nijhoff, The Hague
 86. Sims JT (1993) Environmental soil testing for phosphorus. *J Prod Agric* 6: 501–507
 87. Singh BB and Jones JP (1976) Phosphorus sorption and desorption characteristics of soil as affected by organic residues. *Soil Sci Soc Am J* 40: 389–394
 88. Skinner RJ, Church BM and Kershaw CD (1992) Recent trends in soil pH and nutrient status in England and Wales. *Soil Use and Manag* 8: 16–20
 89. Smith RV (1976) Nutrient budget of the River Main, Co. Antrim. p. 315–339. In: *Agriculture and Water Quality.* Tech Bull 32, Ministry of Agriculture, Fisheries and Food, London, United Kingdom
 90. Summers, RN, Guise NR and Smirk DD (1993) Bauxite residue (Red Mud) increases phosphorus retention in sandy soil catchments in Western Australia. *Fert Res* 34: 85–94
 91. Svendsen LM and Kronvang B (1991) Phosphorus in the Nordic counties: Methods, Bioavailability, Effects and Measures. A report by the National Environmental Research Institute, Denmark. In Danish (English summaries). 201 pp
 92. Taylor AW and Kunishi HM (1971) Phosphate equilibria on stream sediment and soil in a watershed draining an agricultural region. *J Agric Food Chem* 19: 827–831
 93. Tyler DD and Thomas GW (1977) Lysimeter measurements of nitrate and chloride losses from conventional and no-tillage corn. *J Environ Qual* 6: 63–66
 94. Tunney H (1990) A note on a balance sheet approach to estimating the phosphorus fertilizer needs of agriculture. *Irish J Agric Res* 29: 149–154
 95. Uhlen G (1981) Surface runoff and the use of farm manure. pp. 34–43. In Logan JC (ed) *Nitrogen Losses and Surface Runoff from Land Spreading of Manures.* Martinus Nijhoff, The Hague
 96. U.S. Environmental Protection Agency (1990) National water quality inventory. 1988 Report to Congress. Office of Water. U.S. Govt Print Office, Washington, DC
 97. Uunk EJB (1991) Eutrophication of surface waters and the contribution from agriculture. *Proceedings of the Fertilizer Society.* The Fertilizer Society, Peterborough, United Kingdom. 56 pp
 98. Vighi M and Chiaudani G (1987) Eutrophication in Europe: the role of agricultural activities. pp. 213–257. In Hodgson E (ed) *Reviews in Environmental Toxicology* 3. Elsevier
 99. Vincent WF and Downes MT (1980) Variation in nutrient removal from a stream by water cress (*Nasturtium Officinale* R. BR.). *Aquatic Bot* 9: 221–235
 100. Ward JC, O'Connor KF and Wei-Bin G (1990) Phosphorus losses through transfer, runoff and soil erosion. p. 167–183. In: *Phosphorus requirements for sustainable agriculture in Asia and Oceania.* Int Rice Res Inst, Manila, Philippines
 101. White E (1982) Eutrophication in New Zealand lakes. p. 129–136. In: *Water in New Zealand's Future.* Proc. 4th. National Water Conference, August 24–26, 1982, Auckland. Bielecki Printers Ltd., Hamilton, New Zealand
 102. Wilcock RJ (1986) Agricultural run-off: a source of water pollution in New Zealand. *New Zealand J Agric Sci* 20: 98–103
 103. Withers PJA (1993) The Significance of Agriculture as a Source of Phosphorus Pollution in Inland and Coastal Waters in the United Kingdom. A report to the Ministry of Agriculture, Fisheries, and Food, London, United Kingdom. 120 pp
 104. Withers PJA, Unwin RJ, Grylls JP and Kane R (1994) Effects of withholding phosphate and potash fertilizer on grain yield of cereals and on plant-available phosphorus and potassium in calcareous soils. *European J Agron* 3: 1–8
 105. Wolf AM, Baker DE, Pionke HB and Kunishi HM (1985) Soil tests for estimating labile, soluble, and algae-available phosphorus in agricultural soils. *J Environ Qual* 14: 341–348